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TOPIC(s): Alternative solvents / Catalytic systems

Highly Selective Cyclohexane Oxidation Catalyzed by Ferrocene in Ionic Liquid Medium

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PURPOSE OF THE ABSTRACT

The partial oxidation of cyclohexane by aqueous tert-butyl hydroperoxide in phosphonium ionic liquid medium [P6.6.6.14][DCA] and in the presence of catalytic amounts of ferrocene was investigated. The reaction proceeded during 2 h at room temperature with high selectivity (> 98%), yielding mainly cyclohexanone (up to 16%) with total TOFs up to 1 x 104 h-1. The combination of a commercial iron-complex catalyst (ferrocene) and well-adjusted unconventional reaction conditions led to a highly selective, fast and reusable catalytic system for the mild oxidation of cyclohexane. Moreover, the found [Fe(C5H5)2]/ [P6.6.6.14][DCA] catalytic system can be of applied significance to produce polyamide 6.

Therefore, with this work we will discuss the new possibilities in the application of RTILs to selective oxidation catalysis.

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FIGURES



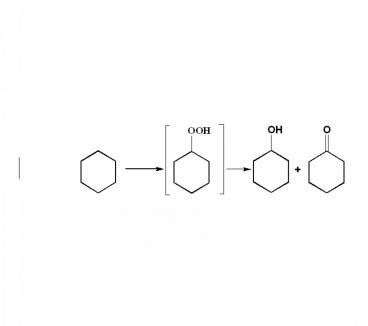


FIGURE 1 Figure 1 Multiphasic mixture obtained by oxidation of cyclohexane with TBHP in [P6.6.6.14][DCA] catalyzed by ferrocene.

FIGURE 2 Figure 2 Cyclohexane oxidation

KEYWORDS

Alkane functionalization | Ferrocene | Ionic liquid | Catalysis

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